

# Analysis of Drugs from Biological Samples

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**Abstract:- Samples which are obtained from the biological sources like human beings are analyzed for various biochemical compounds for screening. these samples are subjected to various methods of preparation to achieve the desired information. However, we need to consider the various parameters of biological samples while preparation of samples. degradation of samples reduced by the checking of physicochemical parameters. there are several types of methodes for estimation and extraction of biochemical compounds like SPE, LLE. as part to know the quantity of the chemical we need to analyses through the HPLC, HPTLC, LC-MS.**

**Keywords:-** SPE, LLE, Biological Samples, LC-MS.

## I. INTRODUCTION

Analysis of drug samples is always essential for the estimation of therapeutical action of the drugs, there are various methods for the analysis in biological samples like direct and indirect methods. in direct, measuring the drug concentration directly through various concentration measurement tests. In indirect methods discussion with the patient and pill courts at alternate intervals throughout the treatment.

In the direct method, assessments are made based on evidence provided by patients or caregivers on presumptive compliance using an electronic medical event monitoring system, and this is based on either qualitative or quantitative measurement of the drug under investigation in a biological matrix provided by the system. A more objective assessment of patient adherence has its own limitations. The most often utilized matrix is plasma; however, the fundamental limitation of this approach is that it only provides the denarian with the drug concentration in the systemic circulation at the time of sampling. This concentration is primarily related on the time interval between providing the last dose and collecting the sample (1-8).

Analysing drugs from biological samples is a critical aspect of forensic toxicology and clinical pharmacology. The process involves extracting, identifying, and quantifying drugs or their metabolites from various biological matrices, such as blood, urine, saliva, hair, and tissue samples. Here's an overview of the steps involved in the analysis:

### ➤ Sample Collection

Biological samples are collected from individuals suspected of drug use or overdose. The type of sample collected depends on factors such as the drug of interest, the window of detection, and the sample availability.

### ➤ Sample Preparation

This step involves processing the collected sample to extract the drugs or their metabolites. Sample preparation methods vary depending on the nature of the sample and the analytical technique to be used. Techniques include liquid-liquid extraction, solid-phase extraction, and protein precipitation.

### ➤ Instrumental Analysis

After sample preparation, the extracted drugs are analyzed using sophisticated analytical instruments such as gas chromatography-mass spectrometry (GC-MS), liquid chromatography-mass spectrometry (LC-MS), high-performance liquid chromatography (HPLC), or immunoassays. These techniques offer high sensitivity, specificity, and selectivity in detecting and quantifying drugs.

### ➤ Identification and Quantification

Once the drugs are separated by the chromatographic system, they are identified based on their retention times and mass spectra compared to reference standards. Quantification is typically performed by comparing the peak areas or heights of the analyte to those of known standards or internal standards added during sample preparation.

➤ *Data Analysis and Interpretation*

The data obtained from instrumental analysis are processed and interpreted to determine the concentration of drugs in the sample. Results are compared to established cutoff concentrations or therapeutic ranges to assess the individual's drug exposure or toxicity.

➤ *Reporting*

The findings of the drug analysis are documented in a comprehensive report, which may include information on the drugs detected, their concentrations, the analytical methods used, and any relevant interpretations or conclusions.

➤ *Quality Control*

Throughout the analysis, quality control measures are implemented to ensure the accuracy, precision, and reliability of the results. This includes the use of calibration standards, quality control samples, instrument calibration, and adherence to established standard operating procedures.

➤ *Expert Testimony*

In legal cases or clinical settings, forensic toxicologists or pharmacologists may be called upon to provide expert testimony based on the results of drug analysis. They may explain the analytical procedures, present the findings, and offer interpretations to support legal or medical decisions.

Overall, the analysis of drugs from biological samples plays a crucial role in various fields, including forensic science, clinical medicine, workplace drug testing, and drug rehabilitation programs. It helps determine drug exposure, assess impairment or toxicity, support legal investigations, and guide medical interventions (9).

Other pharmacokinetic characteristics, however, can have an effect on concentration. Plasma samples should be taken prior to administering the next dosage. Plasma monitoring is a valuable tool for assessing therapy compliance.

Bioanalysis of samples involves the quantitative measurement of drugs, metabolites, biomarkers, or other analytes in biological matrices like blood, plasma, serum, urine, saliva, tissues, or hair. This process is crucial in pharmacokinetic studies, drug development, clinical research, and forensic toxicology. Here's an outline of the bioanalysis process:

➤ *Sample Collection*

Biological samples are collected from subjects or patients following appropriate protocols and ethical considerations. Samples may be collected at various time points to assess drug concentration changes over time (pharmacokinetics) or to monitor drug levels in clinical settings.

➤ *Sample Preparation*

Biological samples typically require preparation before analysis to remove interfering substances and concentrate the analytes of interest. Common sample preparation techniques include protein precipitation, solid-phase extraction, liquid-liquid extraction, and filtration.

➤ *Instrumental Analysis*

Prepared samples are then subjected to instrumental analysis using techniques such as liquid chromatography (LC), gas chromatography (GC), mass spectrometry (MS), or their combinations (LC-MS/MS, GC-MS). These techniques offer high sensitivity, selectivity, and accuracy in quantifying analytes in complex biological matrices.

➤ *Calibration Curve*

Prior to sample analysis, calibration standards of known concentrations are prepared and analyzed to establish a calibration curve. This curve relates the peak area or height of the analyte to its concentration and is used to quantify the analytes in test samples.

➤ *Quality Control*

Quality control samples (e.g., spiked samples at low, medium, and high concentrations) are analyzed alongside test samples to assess the accuracy and precision of the assay. Quality control measures ensure the reliability and validity of the analytical results.

## II. DATA ANALYSIS

Analytical data obtained from the instrument are processed using dedicated software to calculate the concentration of analytes in the samples. Data analysis may involve integration of chromatographic peaks, calculation of concentrations, and validation of results against predefined acceptance criteria.

➤ *Validation*

Bioanalytical methods undergo validation to demonstrate their reliability, accuracy, precision, and robustness according to regulatory guidelines (e.g., FDA, EMA). Validation parameters include specificity, accuracy, precision, linearity, range, and stability.

➤ *Reporting*

Results of the bioanalysis are documented in a comprehensive report, which includes information on the analyte concentrations, sample preparation methods, instrumental parameters, validation results, and any relevant interpretations or conclusions.

➤ *Regulatory Compliance*

In drug development, bioanalytical methods must comply with regulatory requirements to ensure data integrity and patient safety. This involves adherence to guidelines such as GLP (Good Laboratory Practices) and GCP (Good Clinical Practices).

➤ *Interpretation*

Bioanalytical data are interpreted in the context of the study objectives, pharmacokinetic parameters, safety assessments, and regulatory requirements. Interpretation may involve comparing drug concentrations to therapeutic ranges, assessing drug exposure-response relationships, or evaluating drug metabolism and elimination (10).

➤ *Sample Preparation for Analysis:*

Sample preparation is a critical step in the analysis of biological samples, ensuring accurate and reliable results by removing interferences and concentrating the analytes of interest. The specific method of sample preparation depends on factors such as the nature of the analyte, the complexity of the sample matrix, and the analytical technique to be used. Here's an overview of common sample preparation techniques:

➤ **Protein Precipitation:** This technique involves the addition of a precipitating agent (e.g., organic solvent or acid) to the sample to denature proteins and precipitate them out of solution. After centrifugation, the supernatant containing the analytes of interest is collected for analysis. Protein precipitation is simple and widely used for small molecule analysis in plasma, serum, and urine samples.

➤ **Solid-Phase Extraction (SPE):** SPE involves the retention of analytes on a solid sorbent while interfering compounds are washed away. The retained analytes are then eluted from the sorbent using an appropriate solvent for subsequent analysis. SPE cartridges or plates packed with sorbent materials such as reversed-phase, ion-exchange, or mixed-mode resins are commonly used for sample cleanup and pre-concentration.

➤ **Liquid-Liquid Extraction (LLE):** LLE relies on the partitioning of analytes between two immiscible liquid phases (e.g., organic solvent and aqueous solution). After mixing and phase separation, the organic phase containing the analytes is separated and evaporated to dryness before reconstitution in a suitable solvent for analysis. LLE is effective for extracting analytes with different polarities from complex matrices.

➤ **Solid-Phase Microextraction (SPME):** SPME involves the extraction of analytes directly from the sample matrix onto a coated fiber or other solid-phase substrate. The fiber is then desorbed in the analytical instrument for analysis. SPME is a solvent-free extraction method suitable for

volatile and semi-volatile compounds in liquid, gas, or solid samples.

➤ **Derivatization:** Some analytes may require derivatization to improve their chromatographic properties or detection sensitivity. Derivatization reactions involve chemical modification of functional groups to form more volatile or UV-absorbing derivatives. Common derivatization techniques include silylation, acylation, and esterification.

➤ **Filtration:** Filtration is used to remove particulate matter or debris from liquid samples before analysis. Membrane filters with specific pore sizes are employed to retain particles while allowing the passage of analytes. Filtration is essential for preventing blockages in chromatographic systems and ensuring the integrity of analytical instruments.

➤ **Dilution:** In some cases, samples may be diluted with a suitable solvent to bring the analyte concentration within the linear range of the analytical instrument or to reduce matrix effects. Dilution may also be necessary to adjust sample volumes for compatibility with specific analytical techniques or detection limits.

➤ **Sample Reconstitution:** After sample cleanup and concentration, the analytes are reconstituted in a suitable solvent or mobile phase for analysis. The choice of reconstitution solvent depends on the compatibility with the analytical technique and the requirements for chromatographic separation and detection.

Overall, sample preparation is a crucial step in the bioanalytical workflow, ensuring the accuracy, sensitivity, and reliability of analytical results. Proper sample preparation techniques help minimize matrix effects, improve chromatographic performance, and enhance the detection of analytes in complex biological samples (11).

**Solid-Phase Extraction (SPE)** is a widely used sample preparation technique in analytical chemistry for the cleanup, concentration, and isolation of target analytes from complex matrices. It involves the selective retention of analytes on a solid sorbent material while undesired matrix components are washed away. Here's how SPE typically works:

➤ **Selection of Sorbent Material:** SPE cartridges or disks contain a solid-phase sorbent material packed in a column or disk format. The choice of sorbent material depends on the physicochemical properties of the analytes and the sample matrix. Common sorbent materials include silica-based materials (e.g., reversed-phase C18, C8), polymer-based materials (e.g., styrene-divinylbenzene), and specialty sorbents for specific analyte classes (e.g., ion-exchange resins, molecularly imprinted polymers).

- **Conditioning:** Before sample loading, the SPE cartridge or disk is conditioned with a specific solvent or solvent mixture to activate the sorbent and remove any impurities or contaminants. Conditioning ensures optimal retention and recovery of the analytes.
  - **Sample Loading:** The sample is applied to the conditioned SPE sorbent bed either by gravity flow or under vacuum. During sample loading, analytes of interest adsorb onto the sorbent surface through various mechanisms such as hydrophobic interactions, hydrogen bonding, or ion exchange.
  - **Washing:** After sample loading, the SPE cartridge or disk is washed with one or more solvents to remove interfering matrix components while retaining the analytes of interest on the sorbent bed. The wash solvents are carefully chosen to effectively remove matrix components without eluting the analytes.
  - **Elution:** The retained analytes are eluted from the SPE sorbent bed using a solvent or solvent mixture that disrupts the interactions between the analytes and the sorbent surface. Elution solvents are selected based on their ability to efficiently desorb the analytes while maintaining their chemical stability and compatibility with downstream analytical techniques.
  - **Evaporation and Reconstitution:** Following elution, the eluate containing the analytes is evaporated to dryness under reduced pressure or nitrogen gas stream to concentrate the analytes. The dried residue is then reconstituted in a suitable solvent for subsequent analysis by techniques such as chromatography, mass spectrometry, or spectroscopy.
  - **Quality Control:** Throughout the SPE process, quality control measures are implemented to ensure the reproducibility and reliability of the results. This includes monitoring the efficiency of sorbent conditioning, sample loading, washing, and elution steps through the use of internal standards, recovery standards, and spiked control samples.
- SPE offers several advantages, including improved analyte recovery, reduced matrix interference, enhanced detection sensitivity, and compatibility with a wide range of analytical techniques. It is widely used in various applications such as environmental analysis, pharmaceutical analysis, clinical diagnostics, food and beverage testing, and forensic toxicology ( 12-16).
- **Liquid-liquid extraction (LLE)**, also known as solvent extraction or partitioning, is a sample preparation technique used to separate and concentrate analytes of interest from a liquid sample by transferring them from one solvent phase to another. It's based on the principle that different compounds distribute themselves unequally between two immiscible liquid phases, typically an aqueous phase and an organic solvent phase. Here's how liquid-liquid extraction typically works:
  - **Selection of Solvents:** Two immiscible solvents are chosen, typically one polar (aqueous) and one nonpolar (organic). The choice of solvents depends on the analytes' solubility properties and desired selectivity. Common solvents include water, methanol, ethanol, dichloromethane, ethyl acetate, and hexane.
  - **Mixing:** The liquid sample containing the analytes is mixed vigorously with the organic solvent in a separating funnel or other vessel. The vigorous mixing ensures thorough contact between the two phases, facilitating the transfer of analytes from one phase to another.
  - **Phase Separation:** After mixing, the mixture is allowed to stand until the two phases separate into distinct layers based on their densities. During this phase separation, analytes distribute themselves between the aqueous and organic phases according to their solubilities in each solvent. The organic phase, containing the analytes of interest, forms the upper layer, while the aqueous phase forms the lower layer.
  - **Collection of Organic Phase:** The organic phase, containing the extracted analytes, is carefully collected using a separating funnel or other means. The collected organic phase may undergo further processing, such as evaporation or drying, to concentrate the analytes.
  - **Optional Washing:** In some cases, an additional washing step may be performed to remove impurities or residual components from the organic phase. This involves adding a small amount of a selective solvent to the organic phase, mixing, and allowing phase separation to occur again.
  - **Optional Back Extraction:** In cases where analytes are extracted into the organic phase but need to be returned to the aqueous phase for further processing or analysis, a back extraction step may be performed. This involves adding an appropriate aqueous solution to the organic phase, mixing, and allowing the analytes to transfer back into the aqueous phase.

Liquid-liquid extraction offers several advantages, including simplicity, versatility, and the ability to handle large sample volumes. However, it also has limitations, such as the potential for emulsion formation, solvent waste generation, and the requirement for careful optimization of extraction conditions to achieve desired selectivity and efficiency.

This technique finds applications in various fields, including environmental analysis, pharmaceutical analysis, biochemical research, and food and beverage testing, among others. It is often used as a sample preparation step prior to chromatographic or spectroscopic analysis(17-22).

**Biological sample analysis** involves the use of a variety of equipment and instrumentation to extract, separate, identify, and quantify analytes present in biological matrices. The specific equipment used depends on the nature of the analytes, the complexity of the sample matrix, and the analytical techniques employed. Here's an overview of some common equipment used for biological sample analysis:

- **Centrifuges:** Centrifuges are used to separate components of a biological sample based on their density differences. They are essential for tasks such as separating blood cells from plasma or serum, pelleting cellular debris, and isolating organelles or cellular fractions.
- **Analytical Balances:** Analytical balances are used to accurately measure the mass of samples and reagents. They provide precise weight measurements necessary for preparing standards, dilutions, and sample aliquots in analytical procedures.
- **Liquid Handling Systems:** Automated liquid handling systems, such as pipettes, multichannel pipettes, and robotic liquid handlers, are used for precise and reproducible dispensing of liquids during sample preparation, reagent addition, and assay setup.
- **Sample Preparation Equipment:**
  - **Solid-Phase Extraction (SPE) Systems:** SPE systems automate the extraction and purification of analytes from biological samples using solid-phase sorbents.
  - **Liquid-Liquid Extraction (LLE) Systems:** LLE systems automate the partitioning of analytes between immiscible liquid phases for sample cleanup and enrichment.
  - **Homogenizers:** Homogenizers are used to disrupt and homogenize tissue or cell samples for subsequent extraction of intracellular analytes.
- **Chromatography Systems:**
  - **High-Performance Liquid Chromatography (HPLC):** HPLC systems separate and quantify analytes based on their interaction with a stationary phase and a mobile phase under high pressure.
  - **Gas Chromatography (GC):** GC systems separate volatile analytes based on their partitioning between a stationary phase and a gaseous mobile phase.
  - **Ultra-High Performance Liquid Chromatography (UHPLC):** UHPLC systems provide higher resolution, sensitivity, and faster analysis times compared to conventional HPLC.
- **Mass Spectrometers:**
  - **Liquid Chromatography-Mass Spectrometry (LC-MS):** LC-MS systems combine chromatographic separation with mass spectrometric detection for the identification and quantification of analytes in complex samples.
  - **Gas Chromatography-Mass Spectrometry (GC-MS):** GC-MS systems separate and analyze volatile and semi-volatile compounds based on their mass-to-charge ratios.
- **Spectrophotometers:** Spectrophotometers are used to measure the absorbance or fluorescence of analytes in biological samples. UV-Vis spectrophotometers and fluorescence spectrophotometers are commonly used for quantification and detection of specific analytes.
- **Microscopes:** Microscopes are used for visualizing and examining biological samples at the cellular or subcellular level. They are essential for tasks such as histological analysis, cell counting, and morphological studies.
- **Incubators, Shakers, and Temperature Control Equipment:** These equipment are used to provide optimal conditions for sample incubation, mixing, and temperature control during various biological assays and reactions.
- **Data Analysis Software:** Dedicated software packages are used for data acquisition, processing, and analysis in biological sample analysis. These software tools facilitate instrument control, data interpretation, and generation of analytical reports.

Overall, the integration of these equipment and instrumentation enables the comprehensive analysis of biological samples for research, clinical diagnostics, drug development, forensic investigations, and environmental monitoring(23-25).

The ranges of analyte concentrations found in biological samples can vary widely depending on factors such as the type of analyte, the sample matrix, the physiological state of the individual, and the purpose of the analysis. Here are some general ranges for common analytes found in biological samples:

- **Small Molecules (e.g., drugs, metabolites, nutrients):**
  - Microgram per milliliter ( $\mu\text{g/mL}$ ) to milligram per milliliter ( $\text{mg/mL}$ ) for concentrations of drugs and metabolites in blood plasma or serum.



- Nanogram per milliliter (ng/mL) to microgram per milliliter ( $\mu\text{g/mL}$ ) for concentrations of drugs and metabolites in urine.
- Milligram per deciliter (mg/dL) to gram per deciliter (g/dL) for concentrations of glucose, cholesterol, and other small molecules in blood plasma or serum.

➤ *Proteins and Enzymes*

- Milligram per deciliter (mg/dL) to gram per deciliter (g/dL) for concentrations of total protein, albumin, and specific enzymes in blood plasma or serum.
- Microgram per milliliter ( $\mu\text{g/mL}$ ) to milligram per milliliter (mg/mL) for concentrations of specific proteins or antibodies in biological fluids.

➤ *Hormones:*

- Nanogram per milliliter (ng/mL) to microgram per milliliter ( $\mu\text{g/mL}$ ) for concentrations of hormones such as cortisol, testosterone, estrogen, insulin, and thyroid hormones in blood plasma or serum.

➤ *Vitamins and Minerals:*

- Microgram per milliliter ( $\mu\text{g/mL}$ ) to milligram per milliliter (mg/mL) for concentrations of vitamins (e.g., vitamin D, vitamin B12) and minerals (e.g., iron, calcium) in blood plasma or serum.

➤ *Cellular Components:*

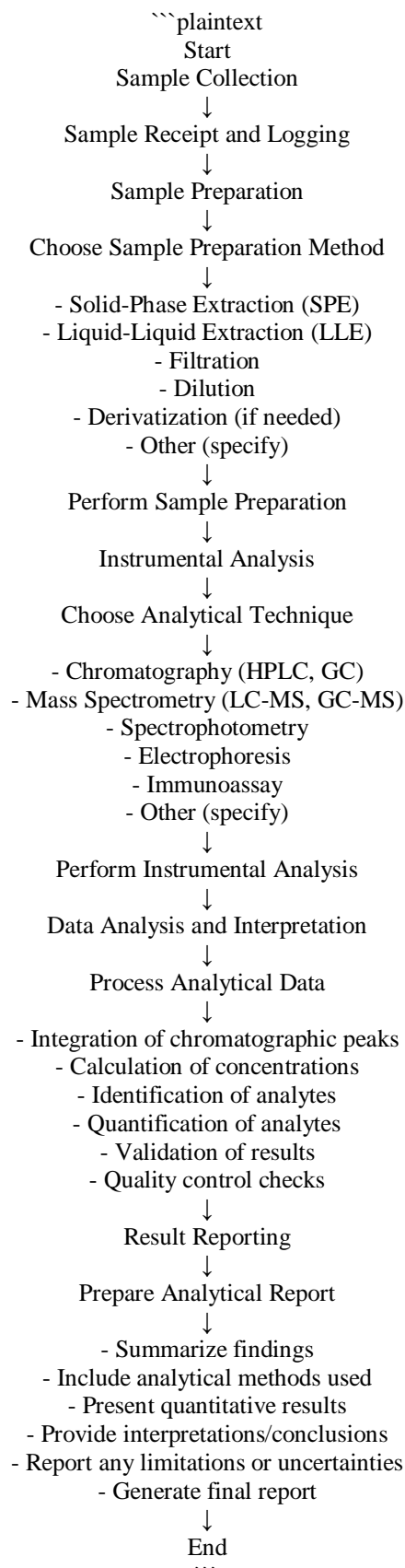
- Cells counts ranging from cells per microliter (cells/ $\mu\text{L}$ ) to cells per milliliter (cells/mL) for white blood cells, red blood cells, platelets, and other cellular components in blood.
- Concentrations of cellular components (e.g., DNA, RNA, proteins) in tissue homogenates or cell lysates, typically measured in microgram per milliliter ( $\mu\text{g/mL}$ ) or nanogram per milliliter (ng/mL).

➤ *Microorganisms and Infectious Agents:*

- Microbial loads measured in colony-forming units per milliliter (CFU/mL) or genome copies per milliliter (copies/mL) for bacteria, viruses, fungi, and parasites in clinical samples (e.g., blood, urine, cerebrospinal fluid).

These ranges are approximate and can vary depending on the specific assay methods, detection limits of the instrumentation, and biological variability among individuals. Quantitative measurements of analytes in biological samples are crucial for assessing health status, diagnosing diseases, monitoring treatment responses, and conducting biomedical research(26-27).

Creating a flowchart for biological sample analysis can help visualize the step-by-step process involved in analysing samples. Here's a basic flowchart outlining the typical workflow:



This flowchart outlines the major steps involved in biological sample analysis, from sample collection and preparation to instrumental analysis, data interpretation, and result reporting. Depending on the specific requirements of the analysis and the nature of the samples, additional steps or variations in the workflow may be included.

### III. PHYSICOCHEMICAL PROPERTIES OF DRUG AND THEIR EXTRACTION FROM BIOLOGICAL MATERIAL

The physicochemical properties of a drug play a crucial role in its extraction from biological materials. Understanding these properties helps in selecting appropriate extraction methods to isolate the drug from complex biological matrices efficiently. Here are some key physicochemical properties of drugs and their implications for extraction:

- **Solubility:** The solubility of a drug in different solvents determines the choice of extraction solvent. Drugs with high solubility in organic solvents are often extracted using liquid-liquid extraction (LLE), while those with higher solubility in aqueous solvents may be extracted using solid-phase extraction (SPE) or other aqueous-based methods.
- **Partition Coefficient (Log P):** The partition coefficient, represented by Log P, indicates the distribution of a drug between a nonpolar solvent (e.g., octanol) and water. Drugs with higher Log P values tend to partition more into organic solvents, making them suitable candidates for extraction using LLE or SPE with nonpolar sorbents.
- **Ionic Character:** Ionic drugs may require specific extraction methods based on their charge. For example, ionizable drugs can be extracted using ion-exchange SPE columns, where the pH of the sample and elution buffers can be adjusted to control the ionization state and enhance extraction efficiency.
- **Molecular Weight and Size:** Large molecular weight drugs or those with bulky structures may exhibit poor diffusion properties, affecting their extraction efficiency. In such cases, mechanical disruption techniques like homogenization or sonication may be used to improve sample extraction.
- **Chemical Stability:** The chemical stability of a drug influences the choice of extraction conditions to prevent degradation during the extraction process. Extraction methods that involve harsh conditions (e.g., high temperature, strong acids or bases) may not be suitable for thermally labile or pH-sensitive drugs.

- **Analyte Concentration Range:** The concentration range of the drug in biological samples dictates the sensitivity requirements of the extraction method and subsequent analytical technique. Sample preparation techniques like SPE or protein precipitation may be needed to concentrate low-abundance drugs for accurate quantification.
- **Matrix Effects:** Biological samples often contain endogenous compounds that can interfere with drug extraction and analysis. Selectivity of the extraction method is crucial to minimize matrix effects and ensure accurate quantification of the drug.
- **Analytical Technique Compatibility:** The choice of extraction method should be compatible with the subsequent analytical technique used for drug quantification. For example, if the drug is to be analyzed by liquid chromatography-mass spectrometry (LC-MS), extraction methods compatible with LC-MS sample preparation, such as SPE, are preferred (28-30).

By considering these physicochemical properties of drugs, researchers and analysts can optimize extraction methods to efficiently isolate drugs from biological materials while maintaining their integrity for accurate quantification and analysis.

Here's a simplified pictorial representation of the extraction process in biological sample analysis, focusing on solid-phase extraction (SPE), one of the commonly used extraction techniques:

#### 1. Step 1: Sample Loading

![Step 1: Sample Loading](https://i.ibb.co/7SWVgL/SPE-Step-1.png)

- The biological sample containing the analytes of interest is loaded onto the SPE cartridge or disk containing a solid-phase sorbent material.

#### 2. Step 2: Sorbent Interaction

![Step 2: Sorbent Interaction](https://i.ibb.co/q9w5pwN/SPE-Step-2.png)

- The analytes of interest selectively interact with the sorbent material based on their physicochemical properties, while interfering matrix components are retained or washed away.

#### 3. Step 3: Washing

![Step 3: Washing](https://i.ibb.co/5BsLkpv/SPE-Step-3.png)

- The SPE cartridge or disk is washed with one or more solvents to remove any remaining matrix components and impurities, ensuring optimal analyte recovery and purity.

#### 4. Step 4: Elution

![Step 4: Elution](https://i.ibb.co/0sK6F1q/SPE-Step-4.png)

- The retained analytes are eluted from the SPE sorbent material using a suitable elution solvent, disrupting the interactions between the analytes and the sorbent to release them into the eluate.

#### 5. Step 5: Concentration

![Step 5: Concentration](https://i.ibb.co/Z6g5N7B/SPE-Step-5.png)

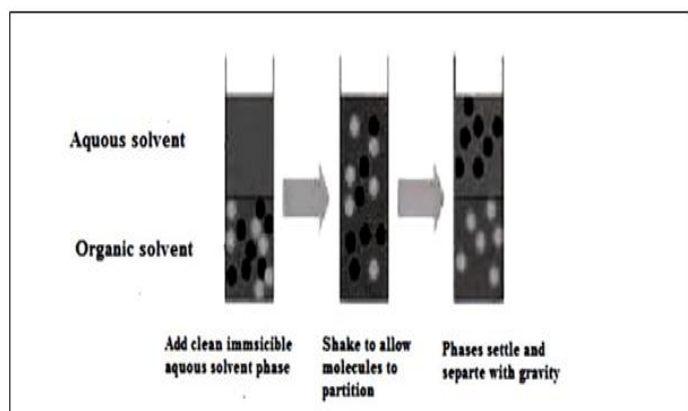
- The eluate containing the analytes of interest is concentrated, often through evaporation or drying, to reduce the volume and increase the analyte concentration for subsequent analysis.

#### 6. Step 6: Reconstitution

![Step 6: Reconstitution](https://i.ibb.co/bQPCJ59/SPE-Step-6.png)

- The concentrated analytes are reconstituted in a suitable solvent or mobile phase for injection into the analytical instrument for quantification and analysis.

This pictorial representation illustrates the key steps involved in solid-phase extraction (SPE) for extracting analytes from biological samples. SPE is widely used for sample cleanup, enrichment, and isolation of analytes prior to analysis by techniques such as chromatography, mass spectrometry, or spectroscopy.



**Fig 1:** Extraction of Biological Samples

#### ➤ Principle Modes of SPME

Direct extraction. In the direct-extraction mode, the coated fiber is placed directly into the sample, and analytes are removed from the sample matrix into the extraction phase. Natural air convections and high diffusion coefficients are usually enough for gaseous samples to equilibrate quickly. For aqueous matrices, more efficient agitation methods are needed, such as forced flow, fast fiber or vial movement, stirring, or sonication.

#### ➤ Headspace Configuration

Headspace mode samples the vapor above the bulk matrix. Thus, analytes must be sufficiently volatile in order to travel from the bulk matrix to the fiber covering.

#### ➤ Advantages

Headspace sampling protects the fiber coating against damage caused by hostile matrices, such as those with extremely high or low pH, or those containing big molecules, such as proteins, which tend to dirty the coating. Biological sample analysis plays a crucial role in various areas of the medical field, contributing to diagnostics, patient monitoring, treatment optimization, and research. Here are some key applications:

#### ➤ Differences between Solid Phase and Liquid Liquid Extraction:

Solid-phase extraction (SPE) and liquid-liquid extraction (LLE) are both sample preparation techniques used to isolate and concentrate analytes from complex mixtures. While they serve similar purposes, they differ in several aspects, including mechanism, selectivity, ease of use, and applicability. Here are the key differences between SPE and LLE:

#### ➤ Mechanism:

- **Solid-Phase Extraction (SPE):** In SPE, analytes are retained on a solid sorbent material (the stationary phase) while interfering matrix components are washed away. The analytes are then eluted from the sorbent using an appropriate solvent. SPE relies on interactions such as adsorption, partitioning, and ion exchange between the analytes and the solid phase.
- **Liquid-Liquid Extraction (LLE):** In LLE, analytes partition between two immiscible liquid phases, typically an organic solvent and an aqueous solution. The analytes distribute themselves between the two phases based on their solubilities in each solvent, allowing for their isolation from the sample matrix.

#### ➤ Selectivity:

- **SPE:** SPE offers high selectivity due to the specific interactions between the analytes and the sorbent material. Different types of sorbents can be selected to target analytes with particular physicochemical properties (e.g., reversed-phase SPE for hydrophobic analytes, ion-exchange SPE for charged analytes).



- LL: LLE selectivity depends primarily on the choice of solvents and their partition coefficients with the analytes. It may be less selective than SPE, especially when dealing with complex sample matrices containing a wide range of compounds.
- Ease of Use:
  - SPE: SPE can be more labor-intensive and time-consuming than LLE due to the need for conditioning, loading, washing, and elution steps. However, SPE cartridges or plates are commercially available with pre-packed sorbent materials, simplifying the process.
  - LLE: LLE is relatively simpler and requires fewer steps than SPE. However, it may involve additional handling of organic solvents and careful manipulation to achieve phase separation, especially with viscous or emulsified samples.
- Applicability:
  - SPE: SPE is versatile and suitable for a wide range of analytes and sample matrices. It is commonly used in environmental analysis, pharmaceutical analysis, clinical diagnostics, and forensic toxicology.
  - LLE: LLE is suitable for isolating analytes with moderate to high partition coefficients between aqueous and organic solvents. It is commonly used in organic synthesis, extraction of natural products, and purification of organic compounds.
- Sample Throughput:
  - SPE: SPE can be automated and adapted for high-throughput applications using robotic systems, allowing for simultaneous processing of multiple samples.
  - LLE: LLE is typically performed manually and may be less amenable to high-throughput processing due to the need for manual phase separation and handling of multiple solvent layers (31-33).

In summary, both SPE and LLE are valuable sample preparation techniques with distinct advantages and limitations. The choice between the two depends on factors such as analyte properties, sample matrix complexity, selectivity requirements, ease of use, and throughput considerations.

#### IV. APPLICATIONS OF BIOLOGICAL SAMPLE ANALYSIS IN MEDICAL FIELD

- Disease Diagnosis: Biological sample analysis enables the detection and diagnosis of various diseases and medical conditions. For example:
  - Blood tests can diagnose conditions such as anemia, infections, and metabolic disorders.
  - Urinalysis helps diagnose kidney diseases, urinary tract infections, and metabolic disorders.
  - Molecular testing of tissues or body fluids can identify genetic disorders, infectious diseases, and cancer.

- Therapeutic Drug Monitoring: Monitoring drug concentrations in biological samples such as blood or plasma helps ensure optimal drug dosing and therapeutic efficacy while minimizing toxicity and adverse effects. Therapeutic drug monitoring is crucial for medications with narrow therapeutic windows, such as certain antibiotics, antiepileptics, and immunosuppressants.
- Pharmacogenomics: Biological sample analysis, including genetic testing, enables personalized medicine approaches by identifying genetic variations that affect an individual's response to medications. Pharmacogenomic testing helps predict drug efficacy, adverse drug reactions, and optimal drug dosing based on an individual's genetic profile.
- Infectious Disease Testing: Analysis of biological samples for the presence of infectious agents, such as bacteria, viruses, fungi, and parasites, aids in the diagnosis and management of infectious diseases. Techniques such as polymerase chain reaction (PCR), serology, and culture are used to detect and identify pathogens in samples like blood, urine, sputum, and cerebrospinal fluid.
- Monitoring Disease Progression and Treatment Response: Serial analysis of biological samples over time allows clinicians to monitor disease progression and evaluate the effectiveness of treatments. Biomarkers present in blood, urine, or other bodily fluids provide valuable insights into disease severity, treatment response, and prognosis.
- Screening and Early Detection: Biological sample analysis is used for population screening programs aimed at early detection of diseases such as cancer, cardiovascular disorders, and metabolic syndromes. Screening tests, including mammography, Pap smears, and blood glucose tests, help identify individuals at risk before symptoms manifest.
- Clinical Research and Drug Development: Biological sample analysis is fundamental in clinical research and drug development processes. It aids in assessing the safety, efficacy, and pharmacokinetics of investigational drugs, as well as in identifying biomarkers for disease diagnosis, prognosis, and therapeutic response.
- Forensic Medicine: Biological sample analysis is utilized in forensic medicine for determining causes of death, identifying toxins or drugs in postmortem samples, establishing paternity, and providing evidence in legal cases such as sexual assault and homicide investigations (34,35).

Overall, biological sample analysis plays a central role in modern medicine, enabling early disease detection, personalized treatment strategies, and advancements in medical research and healthcare delivery.

## V. CONCLUSION

Biological sample analysis is a vital component of modern medical and scientific research, providing invaluable insights into disease diagnosis, treatment monitoring, drug development, and understanding biological processes. Through the meticulous examination of biological samples such as blood, urine, tissues, and cells, researchers and healthcare professionals can uncover a wealth of information about the body's physiological state, the presence of pathogens, the effectiveness of treatments, and genetic predispositions to diseases. Here are some key conclusions about biological sample analysis:

- **Diagnostic Power:** Biological sample analysis enables the accurate diagnosis of a wide range of diseases and medical conditions, from infectious diseases and genetic disorders to cancer and metabolic syndromes. Biomarkers present in biological samples serve as indicators of disease presence, severity, and progression, aiding in early detection and timely intervention.
- **Personalized Medicine:** Advances in biological sample analysis, particularly in genomics and proteomics, have paved the way for personalized medicine approaches. By analyzing an individual's genetic makeup, gene expression patterns, and biomarker profiles, healthcare providers can tailor treatments to each patient's unique characteristics, maximizing therapeutic efficacy and minimizing adverse effects.
- **Therapeutic Monitoring:** Biological sample analysis plays a crucial role in therapeutic drug monitoring, ensuring optimal dosing and therapeutic outcomes while minimizing toxicity and adverse reactions. By measuring drug concentrations in biological fluids, clinicians can adjust treatment regimens based on individual patient responses, improving patient care and treatment outcomes.
- **Research Advancements:** Biological sample analysis fuels scientific discoveries and innovations across various fields, including pharmacology, molecular biology, and translational medicine. By elucidating molecular mechanisms, identifying disease biomarkers, and characterizing drug responses, researchers can develop novel therapeutic strategies, biomarker-based diagnostics, and targeted therapies for improved patient care.
- **Quality Assurance:** Rigorous quality control measures and standardized protocols are essential for ensuring the accuracy, reproducibility, and reliability of biological

sample analysis results. Adherence to regulatory guidelines and proficiency testing programs helps maintain high standards of analytical performance and data integrity, fostering trust and confidence in clinical and research findings.

- **Interdisciplinary Collaboration:** Biological sample analysis requires interdisciplinary collaboration among clinicians, scientists, laboratory technicians, and bioinformatics experts. By combining expertise in medicine, biology, chemistry, and informatics, multidisciplinary teams can tackle complex healthcare challenges, drive innovation, and translate research findings into clinical practice.

In conclusion, biological sample analysis is an indispensable tool for advancing medical knowledge, improving patient care, and addressing global health challenges. Continued advancements in analytical techniques, biomarker discovery, and data interpretation will further enhance the utility and impact of biological sample analysis in healthcare and scientific research.

## REFERENCES

- [1]. Holland NT, Smith MT, Eskenazi B, Bastaki M. Biological sample collection and processing for molecular epidemiological studies. *Mutation Research/Reviews in Mutation Research*. 2003 Jun 1;543(3):217-34.
- [2]. Niu Z, Zhang W, Yu C, Zhang J, Wen Y. Recent advances in biological sample preparation methods coupled with chromatography, spectrometry and electrochemistry analysis techniques. *TrAC Trends in Analytical Chemistry*. 2018 May 1;102:123-46.
- [3]. Wu J, Liu J, Li S, Peng Z, Xiao Z, Wang X, Yan R, Luo J. Detection and analysis of nucleic acid in various biological samples of COVID-19 patients. *Travel medicine and infectious disease*. 2020 Sep 1;37:101673.
- [4]. Morais CL, Lima KM, Singh M, Martin FL. Tutorial: multivariate classification for vibrational spectroscopy in biological samples. *Nature Protocols*. 2020 Jul;15(7):2143-62.
- [5]. Burguera M, Burguera J. Analytical methodology for speciation of arsenic in environmental and biological samples. *Talanta*. 1997 Sep 1;44(9):1581-604.
- [6]. Hoffmann G, Aramaki S, Blum-Hoffmann E, Nyhan WL, Sweetman L. Quantitative analysis for organic acids in biological samples: batch isolation followed by gas chromatographic-mass spectrometric analysis. *Clinical chemistry*. 1989 Apr 1;35(4):587-95.
- [7]. Hernandez F, Sancho JV, Pozo OJ. Critical review of the application of liquid chromatography/mass spectrometry to the determination of pesticide residues in biological samples. *Analytical and bioanalytical chemistry*. 2005 Jun;382:934-46.

- [8]. Szökő É, Tábi T. Analysis of biological samples by capillary electrophoresis with laser induced fluorescence detection. *Journal of pharmaceutical and biomedical analysis*. 2010 Dec 15;53(5):1180-92.
- [9]. Álvarez-Sánchez B, Priego-Capote F, De Castro ML. Metabolomics analysis I. Selection of biological samples and practical aspects preceding sample preparation. *TrAC Trends in Analytical Chemistry*. 2010 Feb 1;29(2):111-9.
- [10]. Niu Z, Zhang W, Yu C, Zhang J, Wen Y. Recent advances in biological sample preparation methods coupled with chromatography, spectrometry and electrochemistry analysis techniques. *TrAC Trends in Analytical Chemistry*. 2018 May 1;102:123-46.
- [11]. Nováková L, Vlčková H. A review of current trends and advances in modern bio-analytical methods: chromatography and sample preparation. *Analytica chimica acta*. 2009 Dec 10;656(1-2):8-35.
- [12]. Poole CF. New trends in solid-phase extraction. *TrAC Trends in Analytical Chemistry*. 2003 Jun 1;22(6):362-73.
- [13]. Buszewski B, Szultka M. Past, present, and future of solid phase extraction: a review. *Critical Reviews in Analytical Chemistry*. 2012 Jul 1;42(3):198-213.
- [14]. Żwir-Ferenc A, Biziuk M. Solid Phase Extraction Technique--Trends, Opportunities and Applications. *Polish Journal of Environmental Studies*. 2006 Sep 1;15(5).
- [15]. Simpson NJ. Solid-phase extraction: principles, techniques, and applications. CRC press; 2000 Mar 15.
- [16]. Berrueta LA, Gallo B, Vicente F. A review of solid phase extraction: basic principles and new developments. *Chromatographia*. 1995 Apr;40:474-83.
- [17]. Hanson C, editor. Recent advances in liquid-liquid extraction.
- [18]. Mazzola PG, Lopes AM, Hasmann FA, Jozala AF, Penna TC, Magalhaes PO, Rangel-Yagui CO, Pessoa Jr A. Liquid-liquid extraction of biomolecules: an overview and update of the main techniques. *Journal of Chemical Technology & Biotechnology: International Research in Process, Environmental & Clean Technology*. 2008 Feb;83(2):143-57.
- [19]. Silvestre CI, Santos JL, Lima JL, Zagatto EA. Liquid-liquid extraction in flow analysis: A critical review. *Analytica chimica acta*. 2009 Oct 12;652(1-2):54-65.
- [20]. Othmer DF, White RE, Trueger E. Liquid-liquid extraction data. *Industrial & Engineering Chemistry*. 1941 Oct;33(10):1240-8.
- [21]. Cantwell FF, Losier M. Liquid-liquid extraction. *InComprehensive analytical chemistry 2002 Jan 1 (Vol. 37, pp. 297-340)*. Elsevier.
- [22]. Kula MR, Kroner KH, Hustedt H. Purification of enzymes by liquid-liquid extraction. *Inreaction engineering 2005 Jul 5 (pp. 73-118)*. Berlin, Heidelberg: Springer Berlin Heidelberg.
- [23]. Colombo G, Clerici M, Garavaglia ME, Giustarini D, Rossi R, Milzani A, Dalle-Donne I. A step-by-step protocol for assaying protein carbonylation in biological samples. *Journal of Chromatography B*. 2016 Apr 15;1019:178-90.
- [24]. Bolte S, Cordelières FP. A guided tour into subcellular colocalization analysis in light microscopy. *Journal of microscopy*. 2006 Dec;224(3):213-32.
- [25]. Niu Z, Zhang W, Yu C, Zhang J, Wen Y. Recent advances in biological sample preparation methods coupled with chromatography, spectrometry and electrochemistry analysis techniques. *TrAC Trends in Analytical Chemistry*. 2018 May 1;102:123-46.
- [26]. Holland NT, Smith MT, Eskenazi B, Bastaki M. Biological sample collection and processing for molecular epidemiological studies. *Mutation Research/Reviews in Mutation Research*. 2003 Jun 1;543(3):217-34.
- [27]. Bryan NS, Grisham MB. Methods to detect nitric oxide and its metabolites in biological samples. *Free radical biology and medicine*. 2007 Sep 1;43(5):645-57.
- [28]. Beata Łabowska M, Michalak I, Detyna J. Methods of extraction, physicochemical properties of alginates and their applications in biomedical field—a review. *Open Chemistry*. 2019 Oct 29;17(1):738-62.
- [29]. Ijardar SP, Singh V, Gardas RL. Revisiting the physicochemical properties and applications of deep eutectic solvents. *Molecules*. 2022 Feb 17;27(4):1368.
- [30]. Jing Y, Yan M, Zhang H, Liu D, Qiu X, Hu B, Zhang D, Zheng Y, Wu L. Effects of Extraction Methods on the Physicochemical Properties and Biological Activities of Polysaccharides from *Polygonatum sibiricum*. *Foods*. 2023 May 22;12(10):2088.
- [31]. Chiron S, Fernandez Alba A, Barcelo D. Comparison of on-line solid-phase disk extraction to liquid-liquid extraction for monitoring selected pesticides in environmental waters. *Environmental science & technology*. 1993 Nov 1;27(12):2352-9.
- [32]. Juhascik MP, Jenkins AJ. Comparison of liquid/liquid and solid-phase extraction for alkaline drugs. *Journal of chromatographic science*. 2009 Aug 1;47(7):553-7.
- [33]. Thorstensen CW, Lode O, Christiansen AL. Development of a solid-phase extraction method for phenoxy acids and bentazone in water and comparison to a liquid-liquid extraction method. *Journal of agricultural and food chemistry*. 2000 Dec 25;48(12):5829-33.
- [34]. Dempsey RJ, Davis DG, Buice RG, Lodder RA. Biological and medical applications of near-infrared spectrometry. *Applied Spectroscopy*. 1996 Feb 1;50(2):18A-34A.
- [35]. Nikalje AP. Nanotechnology and its applications in medicine. *Med chem*. 2015 Mar;5(2):081-9.