

Potential of Liquid Smoke from Palm Kernel Shell as Biopreservative to Tuna(*Thunnus* sp) Fish Protein

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Abstract:- The purpose of this research was to evaluate the role of liquid smoke of oil palm kernel as “biopreservative” on the fish during storage toward chemistry/biochemical or microbiological change at to tuna (*thunnus* sp) fish. This research was carried out in three steps that is: 1) Making liquid smoke with temperature pyrolysis 400°C during 90 minutes and liquid smoke fractionation in rank: F1 : liquid temperature 100°C; F2: 101 - 125°C; F3: 126 – 150°C; F4: 151 – 200°C. 2) Proximate analysis of liquid smoke FO and Fraction F2 to tuna fish that analysed with TVB, Total Bacterium, total protein content . The result indicates that the phenol and acid content was more increase with higher fractionation temperature while the carbonyl content was more reduced with higher fractionation temperature. The TVB value for Tuna fish muscle without soaking higher liquid smoke (38.93 mg N/100g and 36.27 mg N/100g) compared to tuna fish muscle of liquid smoke soaking (F0) that is 13.87 mg N/100g and 11.73 mg N/100g and also soaking with F2 fraction muscle tuna 12.27 mg N/100g. Total bacterium (2.67×10^9 cfu/g and 2.10×10^9 cfu / g) compare without soaking liquid smoke (F0) and F2 fraction that is (8.43×10^5 cfu/g and 6.23×10^5 cfu / g) and (8.57×10^5 cfu/g and 6.50×10^5 cfu/g).

Keywords:- Biopreservative, TVB, .

I. INTRODUCTION

Indonesia is the greatest palm kernel producing countries in the world after Malaysia (Prasetyani and Miranti, 2010). In addition to producing high oil palm kernel as of 10.2 million or 36.32% of total world palm kernel production it also produces waste of palm kernel factory. Commonly, waste of palm kernel factory consist of liquid, solid and gas waste (Wahyono, et al, 2004). Solid waste particularly palm kernel shell has actually benefits. One of them is liquid smoke resulted from pyrolysis process of palm kernel shell that can be used as biopreservative as substitution for chemical preservative and also tar and charcoal.

Liquid smoke is produced with dry distillation process from biomass particle at 150-400°C. The liquid smoke produced contain highly acid and also other dominant compounds such as phenol and carbonyl that together with acid play role as carrier of specific aroma, taste and color. The compound has antimicrobial and antioxidant effects so it is called as biopreservative.

Use of liquid smoke as substitution of traditional smoking has developed fast. Fish smoking is a processing method resulting in yellow or reddish brown smoked fish.

Smoking can preserve fish, due to compound component contained in smoke as biopreservative. According to Pszczola (1995) two main compounds of liquid smoke having bactericidal/bacteriostatic effect is phenol and organic acid. In their combination, both compounds work effectively to hamper microbe growth.

Fish and other fishery products included in food stuff are easy to degrade due to high water content so microorganism is easy to grow naturally and fish contain enzyme that can decompose protein to be putrescine isobutylamine, cadaverine and others causing bad odor and fish fat contain multiple unsaturated fat acid that is easy for oxidation or hydrolysis process resulting in pungent odor. Loose composition of fish cell tissue cause micro be can be used easily as growth media so fresh or processed fishery product including smoked fish stored will undergo physical, chemical, microbiological changes that affect its quality and storage age. (Anonymous, 2001).

However, traditional smoking process done with direct smoke of wood firing has many weaknesses such as inconsistent quality, taste and aroma. In addition, difficulty in controlling its process and companied toxic compound PAH (polycyclic aromatic hydrocarbon) that harm health. Use of liquid smoke as one of fish preservation methods can deal with the weak traditional smoking.

There is harmful and toxic compound in liquid smoke, so liquid smoke should be purified by separating/reducing the compound. One of separation/reduction method is fractionation. Fractionation was done by separating existing component based on difference in boiling point.

This research was done by studying tuna muscle damage process both its red meat and white meat particularly its protein during storing and what is role of palm kernel shell liquid smoke as biopreservative in the tuna muscle during storage. Chemical or biochemical, and microbiological changes during its storage are factor playing important role in determination of duration of acceptable muscle condition. Basic information on chemical and microbiological condition of muscle fish during storage can be uses as reference in doing analysis on shelf life of smoked fish preserved with liquid smoke palm kernel shell.

II. METHODOLOGY

A. Material and tools

Material used is palm kernel solid waste resulted from palm kernel plantation PTPN IV Pabatu, Tebing Tinggi, North

Sumatera. Fish used was from TPI Semarang. Size of fish is same with average weight of 500g/fish.

Tools used for pyrolysis process are pyrolysis equipment including reactor tube, heater, smoke channeling pipe and condenser. Chemical component of liquid smoke was identified using GC-MS (shimadzu QP-5000)

➤ *Liquid smoke preparation and fractionation*

Five kilograms solid waste of dry palm kernel was entered into pyrolysis reactor. Pyrolysis process was done at 400°C for 90 minutes and was fractioned gradually with following criteria: F0: liquid smoke, F1: temperature up to 100°C; F2; temperature 101°C-125°C, F3: temperature 126-150°C, F4 temperature 151-200°C. Chemical composition analysis of liquid smoke and its fractions include.

- Phenol contain analysis, using Senter *et al* (1989) method.
- Carbonyl contain analysis, using Lappin and Clark (1951) method.
- Acid contain analysis, Titration method (anonymous, 1990).

➤ *Study of liquid smoke intervention and fraction F2 (101°C-125°C) as biopreservation on muscle of tuna during storage at room temperature*

Muscle Tuna cleaned from head, bone, stomach contain, tail and other was washed using sterile cold water (sterilized Aqua) then. Its muscle fish is separated and cut into 7cm length and 4.5 width. Each is treated as follows: without soaking with liquid smoke (control), soaking at liquid smoke (FO), and soaking with selected fraction (F2). At 3.3% concentration 30 x thinning it was soaked for 1 hour. Each treatment is stored in sterile contained at room temperature and analyzed every day up to five days. Analysis was done on:

- Total volatile based (TVB) (Apriyanto *et al*, 1989).
- Total Bacteria (APC) with Agar method (Post, 1988).
- Total protein level (anonymous, 1990).

III. RESULT AND DISCUSSION

A. Pyrolysis result

Palm kernel shell pyrolysis at 400°C resulted in 2500 liquid smoke component (while its charcoal was 1500g and tar 430 ml).

component	quantity
liquid smoke	2500 ml
charcoal	1500g
tar	430 ml

Table 3.1 Component of palm kernel liquid smoke (5 kg shell weight)

Result obtained in table 3.1 is caused by palm kernel shell used in the research has optimal water content (13%) with temperature and time determined in pyrolysis process so its heat conductivity is greater. Pyrolysis process is perfect so resulted in most liquid smoke and tar, and pyrolysis resulted in relatively little charcoal. Maga (1988) and Girrard (1992) suggest that pyrolysis process involve reaction process of decomposition, oxidation, depolymerization and condensation. Processes of water elimination at 120°C-150°C,

hemicellulose pyrolysis at 200°C -2500c, cellulose pyrolysis at 280-320°C and lignin pyrolysis at 400°C result in compound having high organoleptic quality compound and higher temperature will cause condensation reaction that form new compound and condensation product oxidation was followed with increase in tar compound and polycyclic aromatic hydrocarbon.

Effort to eliminate or reduce PAH component of benzo(a)pyrene in liquid smoke by minimizing accompanied tar compound in liquid smoke by redistillation liquid smoke resulted on temperature that can evaporate main component of liquid smoke (organic acid, phenol and carbonyl) but not evaporate tar compound so resulted in total redistillationtar free liquid smoke. According to Gorbatov *et al* (1991) one of fractionation methods is liquid smoke redistillation to eliminate tar compound and PAH.

➤ *Analysis of liquid smoke chemical composition and its fraction*

Phenol compound content of palm kernel shell liquid smoke and its fraction in figure 3.1

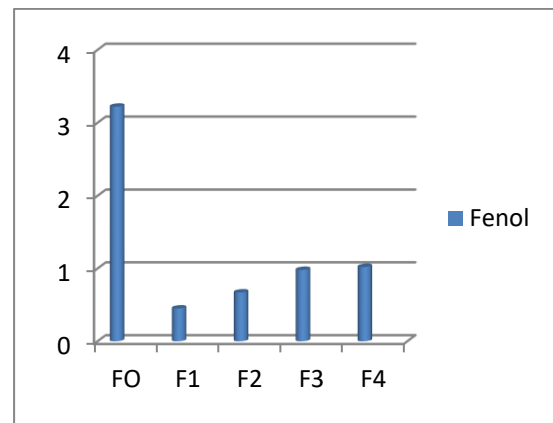


Fig 3.1:- Phenol compound content of palm kernel shell liquid smoke and its fraction

Result of phenol analysis indicate that F0 result in the most phenol (F0 =3.22%, F1=0.45%, F2=0.67%, F3=0.98%, F4=1.02%). Result of fractionation indicated that its phenol contain is lower than liquid smoke. It is caused by phenol compound has higher boiling point and fractionation result using low temperature so resulted in lower phenol from control liquid smoke (FO). However, there is tendency its phenol contain increase along with increase fractionation temperature. Phenol level on FO liquid smoke is due to FO that is total phenol from its fractionation result. According to Buckingham (1982) phenol boiling point was 162°C-285°C so the higher fractionation temperature the higher its phenol. Carbonyl compound contain from palm kernel shell liquid smoke and its fraction at figure 3.2

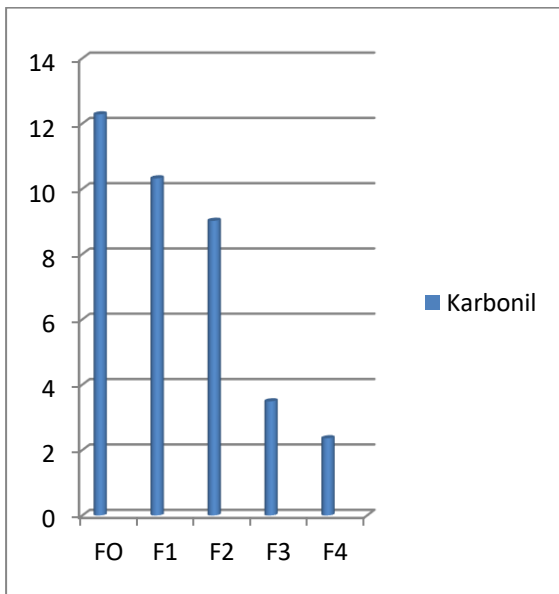


Fig 3.2:- Carbonyl compound contain from palm kernel shell liquid smoke and its fraction

Result of carbonyl test from palm kernel shell liquid smoke indicated its content of 12.28%. Cellulose content on palm kernel shell is lower (32.93%) than cellulose content of coconut shell (38.98%) so carbonyl content of palm kernel shell liquid smoke is higher due to cellulose content at the material will result in carbonyl. High cellulose content will result in high carbonyl. At palm kernel shell liquid smoke carbonyl content is greater than phenol. According to Girrard (1992) phenol has carbonyl group that is considered as carbonyl.

For its fractionation result, carbonyl level is lower than liquid smoke (FO) (12.28%). The great carbonyl content on liquid smoke is influenced by lignin content, cellulose in oil-palm shell due to carbonyl compound is result of fractionation of the component and is raw material having high cellulose content will result in high carbonyl content due to high heating treatment in cellulose will result in carbonyl. Results of fractionation at F1, F2, F3, F4 indicated that higher temperature result in reduced carbonyl content. It is due to boiling point of carbonyl is the lowest among three main components of liquid smoke (less than 100°C). According to Buckingham (1982) in general carbonyl has boiling point less than 100°C. Acid compound content of palm kernel shell liquid smoke and its fraction at figure 3.3

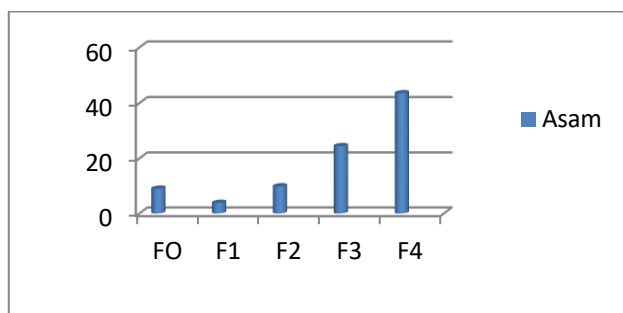


Fig 3.3:- Acid compound content of palm kernel shell liquid smoke and its fraction

Result of fractionation acid level analysis indicated that result of fraction F1 is lower than F2, F3 and F4 (3.83%; 9.83%; 24;43%; 43.51%). It is due to acid compound of liquid smoke consisting of short, moderate and long chain acid compound. Higher temperature results in cut of moderate and long chain acid compound. So, higher temperature will cause greater acid content. Acid content in oil-palm shell liquid smoke (FO) of 8.99% was lower than acid content of coconut shell liquid smoke of 11.39%. The difference is cause by difference in cellulose content at both material. Cellulose content of oil-palm shell and coconut shell were 32.93% and 38.98%, respectively. Its content of oil-palm shell is lower than coconut shell because acid compound is result of cellulose component fractionation. According to Maga (1998) cellulose pyrolysis result in acetate acid compound and carbonyl compound such as acetaldehyde, glyoxal and acrolein. Then Girrard (1992) suggested that cellulose content in material will determine acid, furan, phenol and water content in the liquid smoke produced. It is estimated that cellulose in oil-palm shell liquid smoke (FO) result in high carbonyl content and low acid compound (8.99%).

Benzo(a)pyrene content analysis using GC-MS of palm kernel shell liquid smoke sample was 15.82 ppm while benzo(a)pyrene at fractionation F2 at 101°C-125°C was 0 ppm. According to Sax and Lewis (1987) benzo(a)pyrene is poly cyclicaromatic hydrocarbon (PAH) compound group with five benzene rings found in coal tar, cigarette smoke and atmosphere as imperfect burning. Further, benzo(a)pyrene is yellow crystal with melting point of 179°C and boiling point 310°C-312°C. It agrees to result of research where fractionation used is F2 at 101°C-125°C that resulted in 0 ppm benzo(a) pyrene that is safely used as bio preservative at tuna. Acceptable maximal level of benzo(a) pyrene in food is 10 ppb (10µg/kg), so it is necessary to look for save limit for liquid smoke usage in food by doing thinning. In this research, thinning was done 30 times and 1 hour soaking time so liquid smoke is safe for use in tuna.

➤ Study of liquid smoke intervention and fraction F2 (temperature 101°C-125°C) as biopreservative in tuna during storage at room temperature

• Total Volatile Bases (TVB)

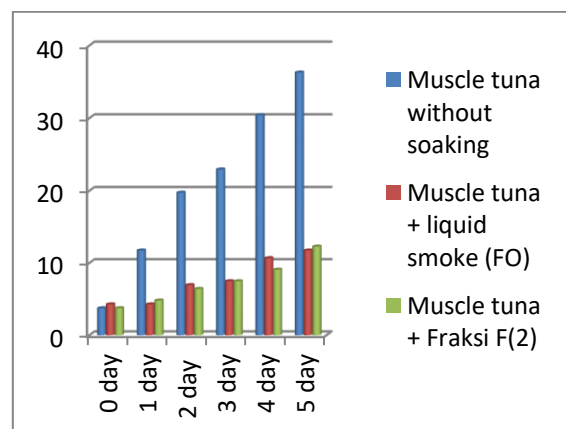


Fig 4.1:- Histogram of volatile base score of tuna without soaking, soaking with liquid smoke (FO), and soaked with fraction F2

At figure 4.1. TVB value for tuna without soaking with liquid smoke indicated increase in TVB during storage. At 0 day and 5 days storage TVB scores are 3.73 mg N/100g and 36.27 mg N/100g. Tuna with soaking at palm kernel shell liquid smoke (FO) can be still consumed because TVB until day 5 was 11.73 mg N/100g. It is due to liquid smoke has smoke component that play role as antibacterial and antioxidant including phenol, and acid that can hamper microbe growth that can react directly with cell membrane causing increase in cell membrane permeability causing lost cell content, inactivation of essential enzymes and genetic material function inactivation. The higher phenol concentration will precipitate microbe protein effectively (Estrada et al, 1998).

TVB value of tuna soaked at liquid smoke fraction F2 from 0-5 day storage indicate hamper of damage due to F2 contain acetic acid and phenol component functioning as antibacterial agent causing bacteria's loss survivability so soaking with fraction 2 liquid smoke stored until 5 days is consumable because its TVB is only 12.27 mg N/100g. However, at day 5 TVB increase due to liquid smoke (FO) or fraction F2 liquid smoke with 30 time thinning. So, storing till 5 days indicate decrease in inactivation by smoke component over microorganism decomposing protein or other compound containing nitrogen with increase base-base volatile resulted from bacteria decomposition. According to Darmadji (1996) acid and phenol improve antibacterial activity. Then, Swastawati *et al* (2007) suggested that phenolic compound has oxidative effect and antibacterial effect that is very effective used in preserving fish because it can add shelf life of fish more than 4 days.

Result of statistic test of tuna meat without soaking indicated change in TVB and significantly different $P \leq 0.01$ with tuna soaked in palm kernel shell liquid smoke (FO) and that soaked with fraction F2 liquid smoke.

Further analysis with Duncan test of tuna without soaking with liquid smoke is significantly different ($P \leq 0.05$) from tuna soaked with liquid smoke (FO) and soaked with F2 liquid smoke.

- *Total bacteria (cfu/g)*

Total bacteria (cfu/g) of muscle tuna without soaking in palm kernel shell liquid smoke increase during storage 0 day and 5 day are 3.47×10^5 (cfu/g) and 2.10×10^9 (cfu/g). According to Frazier and Westhoff (1983) the fish indicate degradation when total bacteria more than 10^6 cfu/g. Analysis on total bacteria at day 2 indicate that tuna without soaking with liquid smoke have degradation and is not consumable. Total bacteria of tuna soaked at liquid smoke (FO) during 0-5 day storage indicate antibacterial activity, so until day 5 total bacteria was 6.23×10^5 cfu/g in which the tuna is consumable. At fraction F2 at 0-4 day storage degradation hamper occur because component playing role as antibacterial is phenol group and acid particularly acetic acid having boiling point of 118°C . There is no increase in bacterial amount at tuna in 0-4 day indicate effect of fraction F2 liquid smoke residue playing role in hampering bacteria. Meanwhile at day 5 total bacteria increase to 6.50×10^5 cfu/g, but it is still consumable.

Decreased hamper at day 5 is due to acetic acid of 9.83% that is antimicrobial compound in F2 liquid smoke. Antimicrobial in liquid smoke decreased because it used by bacteria, so antibacterial activity of fraction F2 liquid smoke will decrease causing increase in total bacteria at day 5 storage. It agrees to Orskow (1984) in which higher resistance of *Escherichia coli* compared with three other bacteria is due to the bacteria may use acetic acid as single carbon source for its life.

Statistical analysis on amount of bacteria in tuna without liquid smoke soaking, soaking with liquid smoke (FO) and soaking with fraction F2 liquid smoke indicated significant difference ($P \leq 0.01$)

Further analysis with Duncan test of tuna without soaking with liquid smoke is significantly different ($P \leq 0.05$) from tuna soaked with liquid smoke (FO) and soaked with F2 liquid smoke.

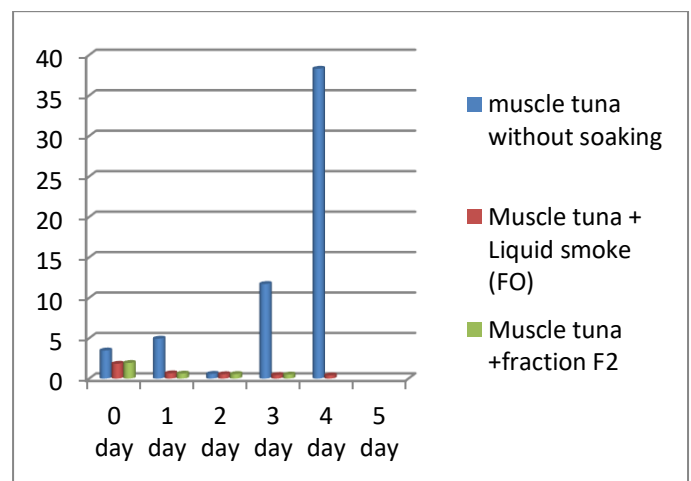


Fig 4.2:- Histogram of total bacteria of tuna without soaking, soaking with liquid smoke (FO), and soaked with F2

IV. CONCLUSION

Based on results of the research the following conclusions are drawn:

- Palm kernel shell liquid smoke and fractionation F2 result can play role as bio preservative for tuna fish.
- Capability of palm kernel shell liquid smoke to hamper protein degradation of tuna affect TVB, bacteria total at storage until 5 day.
- Increase in TVB of tuna without liquid smoke soaking, with liquid smoke (FO) soaking and fraction F2 are 36.27 mg N/100g, 11.73 mg N/100g, and 12.27 mg N/100g, respectively.
- Increase in total bacteria of tuna without liquid smoke soaking, soaking with liquid smoke (FO) and fraction F2 are 1.10×10^9 cfu/g, 6.23×10^5 cfu/g, and 6.50×10^5 cfu/g.
- Phenolic content increase along with increase in fractionation temperature.
- Carbonyl content decrease with increase in fractionation temperature.
- Acid content increase along with increase in fractionation temperature.

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